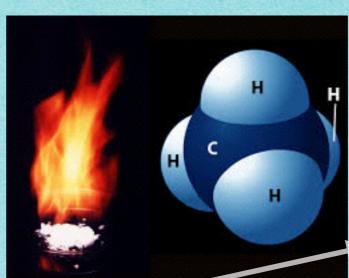
Boyle's Law

Focus: Methane (gas), which is a green house gas, is trapped in our ocean floor as a methane hydrate (solid).

Intro to Methane (CH4)

- Methane is greenhouse gas. What is methane gas?
- Methane molecules are dissolved in our ocean water.
- Methane hydrate is a methane in a solid state at the bottom of shallow water depths.
- Methane hydrates in the ocean floor fill pores, kind of like cement.
- If the hydrates disappear, the shallow ocean floor becomes unstable.

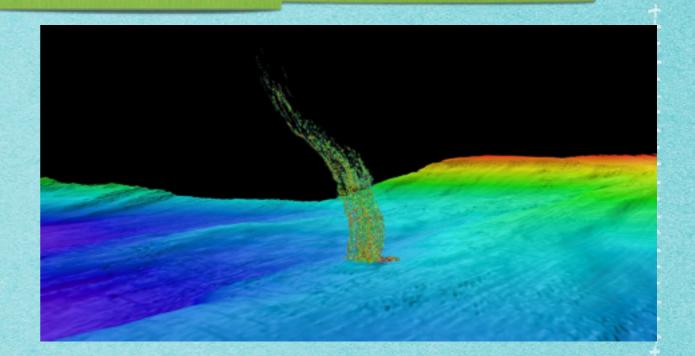






Our current ocean condition

- The ocean temperature is rising due to global warming.
- CO2 is absorbed into our oceans, which is causing it to become more acidic.
- Natural Methane Hydrate
 (solid on bottom of shallow
 ocean floor) is "sublimating"
 because of warmer ocean
 temperatures.
- Methane hydrates are disappearing rapidly in Washington and in the Arctic.



http://
www.popsci.com/
methane-plumes-offwest-coast-may-becaused-by-warmingoceans

Problem with methane hydrate in our warming oceans.

- If methane is released rapidly from the hydrates, it could rise in the form of gas bubbles.
- When this happens at shallow water depths where bubbles can't be absorbed by water.
- Methane hydrates are under low pressure in the shallow ocean shelf of the arctic area.
- Permafrost is melting, also.





effect of methane hydrates disappearing.

- fatal consequences:
- methane hydrate acts as a cement as it fills pores, so as it disappears, the ocean floor becomes unstable, causing under water landslides, which can result in tsunamis.

Also, pressure decreases as gas bubbles are rising. According to Boyle's Law, these bubbles will then increase in volume.

Large bubbles rising to ocean surface changes the density of the water, and the buoyancy of ships: causing ships to sink

videos of methane hydrate in ocean

- https://www.youtube.com/
 watch?v=MSmAXp_BHcQ
- https://www.youtube.com/
 watch?v=iJQzyBqZeZc
- https://www.youtube.com/
 watch?v=OZEXv5YwmgY

